



The Adsorption Mechanism of Activated Carbon and Its Application - A Review

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Abstract. Activated carbon (AC) was recognized by many researchers as useful substance in adsorption of impurities. Several processes involved in the production of AC which were carbonization, crushing, and activation process. Carbonization of carbon required high temperature up to 900°C. Then the carbon will be crush to a desired size for activation process. Activation of carbon can be either chemical activation, physical activation or combination of chemical and physical activation which called physiochemical activation. The mechanism adsorption of AC commonly due to its micropore present in the carbon or the weak vander waals forces which can attract the impurities. Activated carbon have multiple function in human daily life. This study will be discussing the function of AC in the production face mask, water filtration and air filtration.

Keywords: activated carbon; carbonization; vander waals forces; face mask; water filtration; air filtration

DOI: [10.37869/ijatec.v1i3.37](https://doi.org/10.37869/ijatec.v1i3.37)

Received 14 December 2020; Accepted 1 January 2021; Available online 1 January 2021

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1. Introduction

Its was known to world that the usage and application of activated carbon as multipurpose adsorbent. Some using activated carbon as air purification and other used as waster waste treatment. The previous works explained and supported the usage of activated carbon which mentioned that Activated carbon can be call as “material of the future” due to its versatility usage from medical aspect, water and air treatment and a very good adsorbent characteristic [1]. The used of carbon in daily life were back dated in early civilization. [2,3] mentioned that The origin of activated carbon (AC) is associated with Ancient Egypt (1500 BC), whereby the Egyptians make use of its adsorbent characteristics for water purifications and medicinal purposes also highlight that, in olden days (1500 B.C) activated charcoal has been used for medicinal application Before the carbon can be produce and used, the selection of material, process and end product must be identified for a better knowledge regarding the adsorption mechanism of an activated carbon.

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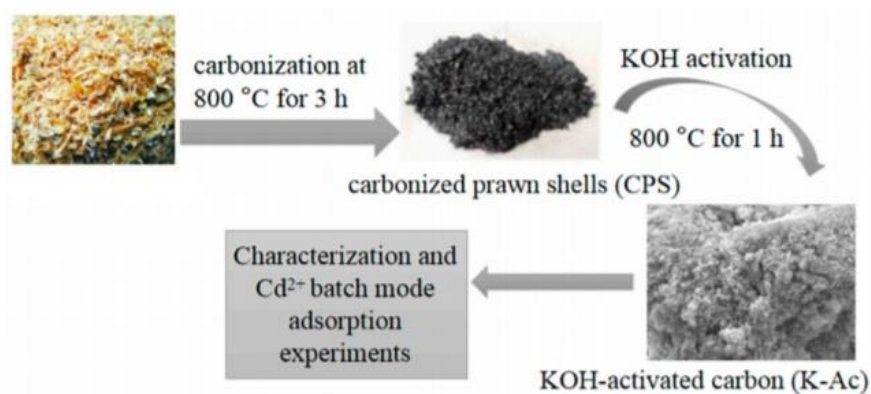


Figure 1. Flow process of AC production [4]

2. Mechanism of Adsorption

The process by which a solid surface concentrates fluid molecule by physical forces is known as adsorption (whereas absorption is a process whereby fluid molecules are taken up by a liquid and distributed throughout that liquid). The adsorption of AC influence by several factor, in this research the adsorption mechanism of AC will be focus on its microspore and weak van der waals forces.

2.1 Microspore structure

Microspore formation in AC formed during the burning process of AC. According to [5], Activated carbons (ACs) are well-known conventional adsorbents with many applications due to their hydrophobicity, surface functionality, pore structure, and high surface area. The raw materials that undergoes burning process will be converted to charcoal. In this process the particle structure of the raw material will be restructure. Activated carbon (AC) was prepared from date palm leaflets using KOH activation followed by nitric acid oxidation to produce oxidized activated carbon (OAC) which possesses acidic and increase in pore number [6]. The bond between each particle will be break and change into new bond which will create the porous of the material. However, the present spore does not effective as the activated carbon since the number of microspores is less. [7] highlighted from study which mentioned that the sorbent is mainly microporous (with a micropore volume equal to 0.31 cm³ g⁻¹) Hence in order to increase the effectiveness the charcoal must be further process which was the activation process. Throughout activation process more and more of the microspore will be formed alternatively increase the surface area of AC. Microspore were very small and only be seen under SEM. Relatively the microspore act similar to sponge. Sponge which has numerus hole in it can traps dust and other substance. The function of microspore in AC can be related to sponge as its allowed substance to be traps inside it in much smaller size.

2.2 Weak van der waals forces

The elemental force causing physical adsorption on activated carbon is the London dispersion force, a form of Van der Waals force, resulting from intermolecular attraction. In the case of adsorption, carbon and the adsorbate are thus chemically unchanged. However, in the process known as chemisorption, molecules chemically react with the carbon's surface (or an impregnant on the carbon's surface) and are held by chemical bonds that are much stronger forces compared to London dispersion forces. The London dispersion force is an intermolecular interaction that exists between all molecules (both polar and non-polar), but it is extremely short-ranged. [8] state that the rich AC has a characteristic that may be favorable for the adsorption of cationic species such as dyes through electrostatic interaction. MB is chosen a typical cationic dye to investigate the adsorption capacity. It is responsible for condensation of most gases to liquids, and the reason higher-molecular-weight gases have higher boiling points. The observed London force is the sum of all the individual interactions of the adsorbate molecule and the neighboring graphite plates composing the carbon structure. The magnitude of the adsorption force will be related to the number of carbon plates, or density of carbon, within the vicinity of the adsorbate molecules. London forces exist between all molecules. Therefore, all molecules adsorb on activated carbon to some extent, depending on their vapor pressure and solubility at the carbon temperature. The London forces are unaffected by

temperature, and thus the adsorption force field will be constant with temperature. The carbon adsorption capacities will still be sensitive, however, to the changes with temperature in vapour pressure or solubility of the adsorbing molecules. The magnitude of the London force is very sensitive to the separation of the adsorbate molecule from the graphite plate. The London force can be considered negligible with a separation greater than about two molecular layers. Therefore, the adsorption forces will only be significant if the gaps or voids within the carbon structure (pore widths) are less than four or five molecular layers. The first three characteristics of London forces are shared by another familiar force: gravity. London forces, and consequently the carbon adsorption forces, are analogous to gravitational forces. However, London forces are much shorter-ranged, and operate on a molecular scale instead of an astronomical scale [9]. [10] support the statement by mentioned that adsorption is the process by which fluid molecules become attached to a surface by physical or chemical forces or combination of both forces. In physical adsorption the impurities are held on the surface of carbon by low level van der Waals forces while in chemisorption the force are relatively strong and occur at active site on the surface.

3. Application of Activated Carbon

Wide range in application of AC drive researcher from all around the world to study more on this useful material. For the purposed of this study, the usage of AC in face mask, air filter and water treatment will be further discussed.

3.1 Face mask

Granular activated carbon (GAC) has been the most common adsorbent used in respirators for protection against gas phase contaminants [11]. Activated carbon fiber (ACF) is considered as an alternative adsorbent to GAC since it has been shown to be effective in capturing volatile organic compounds (VOCs) from gas streams at wide range of concentrations [12,13]. The small fiber diameter of ACF (ranging from 10–20 μm) allows homogeneous activation of the fibers, resulting to a unimodal, narrow pore size distribution in the micropore range. ACFs have been assessed for air pollution control of VOCs Other ACF applications include energy saving and storage, CO₂ capture, and SO₂ removal [14]. Respirator or face mask were important especially in workplace to prevent dust and any substance from entering the lung which can cause sickness.

Table 1. substance present in workplace [15]

No.	Form	Properties	Sample of substance
1.	Solid particle	Particle in solid material, including aerosol, dust, smoke, fibre and fume	Smoke, welding fume, wood dust, engine exhaust, asbestos dust and flour
2.	Liquid particle	Mist, fine spray and aerosol that made a small droplet of liquid	Pesticide, paints, Liquid jetting and powder coating mix
3.	Vapour	Gaseous forms from solid and liquid	Mercury and solvent vapour
4.	Gas	Gaseous	Engine exhaust gases, carbon monoxide, chlorine and sewer gas.

Wearing a respiratory mask is considered to be one of the most affordable and effective methods to reduce exposure to airborne pollutants. For various commercially available respirators, N95 FFRs, surgical mask and activated carbon mask are widely studied [16]. [17] discover that carbon impregnated in fibre cause substrate to attached to elementary fibres but not embedded in their structure, which is important because the deposition of carbon adsorbent in the nonwoven structure can significantly impact the sorption properties of the material. The activated carbon combination mask model group has a higher ability than other mask models in absorbing VOCs [18].

3.2 Air filter

[19] mentioned that Activate carbon can clean the environment by absorbing the harmful gasses (CO_x, NO_x, and SO_x) from the air. These pollutant gases cause respiratory diseases, lung infection,

breathing problems, cough, heart disease, and paralysis [20]. The removal efficiency depends on the characteristics of the feedstock (petroleum coke, coal, biomass), activators (potassium hydroxide, carbon dioxide, and phosphoric acid), and process variables (activation temperature, the ratio of activator to feedstock) [21]. AC filter demand increase vigorously in recent year recent years as a promising alternative adsorbent. They are prepared from fabric precursors and have a very high surface area and are used in the form of cloth or felt, making them much easier to handle than GAC therefore, they should be adaptable to many air cleanings purposes [22]. [23] has develop an integrated carbon fiber brush charger combined with a metallic collection rod and an activated carbon fiber (ACF) sheet that called electrostatic precipitation (ESP).

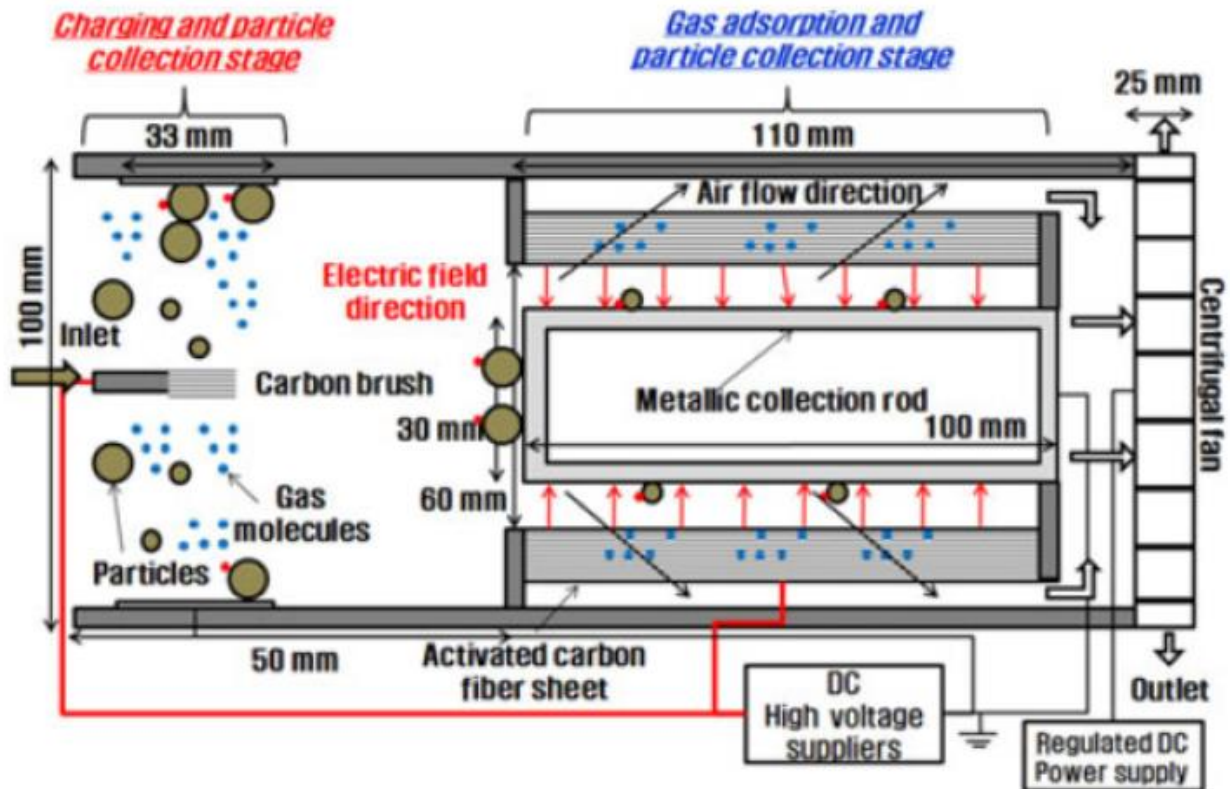


Figure 2. Schematic of the electrostatic precipitation (ESP) type air cleaner [23]

Granular activated carbon (GAC) filters have been widely used to remove harmful gaseous pollutants however, GACs have many weaknesses, including large pressure drops caused by packed media and continuous depletion of granular material. ACF filters are therefore beginning to receive attention as an alternative gaseous pollutant adsorbent [24] and as a gas adsorbent in novel small air purifiers. An excessive exposure to indoor toxic pollutant gases such as trimethylamine (TMA) and H₂S is fatal to humans especially in the confined workspace. A porous carbon pollutant filter was developed using rice husk as a biomass waste. The different carbonization temperature highly affected on the surface area of ARH (activated rice husk carbon) and their performance of pollutant gas adsorption [25].

3.3 Water treatment

Adsorption onto granular activated carbon (GAC) is an established technology in water and advanced wastewater treatment for the removal of organic substances from the liquid phase. Besides adsorption, the removal of particulate matter by filtration and biodegradation of organic substances in GAC contactors has frequently been reported. The application of GAC as both adsorbent for organic micropollutant (OMP) removal and filter medium for solids retention in tertiary wastewater filtration represents an energy- and space saving option, but has rarely been considered because high dissolved organic carbon (DOC) and suspended solids concentrations in the influent of the GAC absorber put a significant burden on this integrated treatment step and might result in frequent

backwashing and unsatisfactory filtration efficiency [26],[27] mentioned that, Among the modifications and methods proposed to improve the performance of water treatment process, the BAC treatment is one of the most promising, environmentally friendly and economically feasible processes. The BAC can overcome several limitations of AC treatment and other conventional water treatment processes. In recent years, there has been growing concern worldwide regarding nitrate (NO_3^-) contamination in groundwater and associated health problems. Most of conducted research have shown that the presence of large nitrate concentrations in potable water resources can potentially lead to the serious troubles, including eutrophication phenomenon and infectious diseases, such as cyanosis and cancer of the alimentary canal [28] and blue-baby syndrome among infants [29]. The World Health Organization (WHO) and US EPA have established the maximum contaminant level (MCL) of 10 mg/L for $\text{NO}_3\text{-N}$ in drinking water. Therefore, the removal of large nitrate concentrations from solution before their discharge into the drinking water resources is so crucial. To date, lots of environmental remediation techniques such as ion exchange, electrodialysis, reverse osmosis, catalytic denitrification and biological denitrification, chemical reductions using zero-valent iron (ZVI), and adsorption [30,31] were developed to remove nitrate from water. Research towards more cost effective without using any chemical in wastewater treatment proved that AC acts as very good adsorbent in filtering the water [32]. ACs are proven to be effective in removal of various pollutants from aqueous solutions, including dyes, pharmaceutical and personal care products (PPCPs), heavy metals, organic pollutants. Thus, the latest research papers on applications of ACs derived from biowaste are grouped into different categories according to the types of adsorbates studied. Each subsection contains a table that summarizes the important aspects of the latest research works, including type of biowaste used, preparation steps of the AC, together with removal efficiency or percentage removal of target pollutant [33]. According to [34], The last treatment of a drinking water treatment plant (DWTP) before water distribution is usually the disinfection process, in which a widely used oxidant is chlorine dioxide (ClO_2). Activated carbon filters are used for dichlorination and adsorption. Dichlorination is a very rapid process, in which the activated carbon participates in the reaction and the free chlorine is converted to chloride.

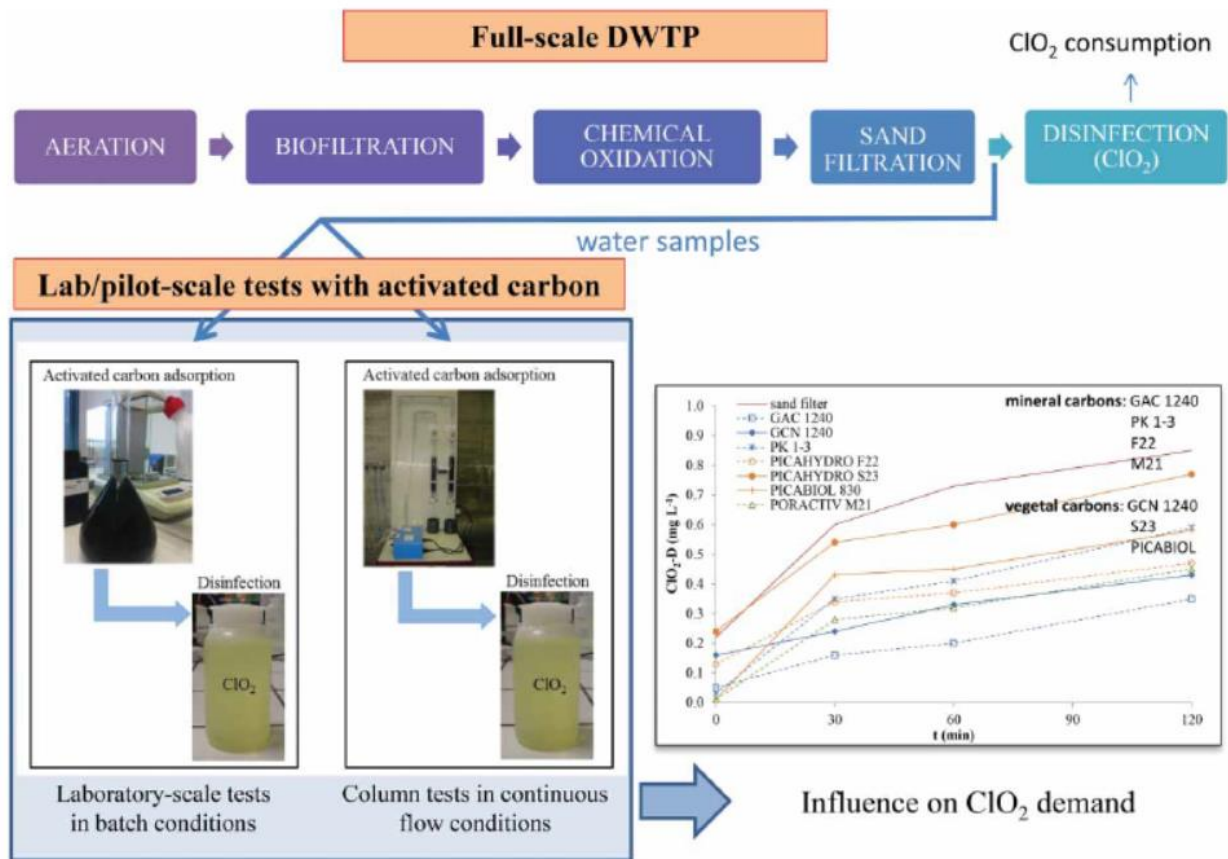


Figure 3. Pilot testing for chlorine oxide remover [34]

4. Conclusion

This study is helpful in enhancing the knowledge regarding the adsorption mechanism of AC. AC were used since the beginning of human civilization as water filtration and medicine. Research on AC in term of perfecting the AC production was done by many researchers according to desire output. Based on this study:

- The materials selection, process involved, and end product of AC can be determined.
- The adsorption of AC which focusing on microspore and van der waals forces can be further studied.
- The application of AC as face mask, air filter and water treatment can be compared thoroughly in the present market demand.

Acknowledgment

The authors acknowledge Faculty of Engineering Technology, Universiti Tun Hussein Onn Malaysia for the lab and facilities provided throughout these studies. Also great thanks to all that contribute to this study directly or indirectly.

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